Chemistry 11 Classification of Matter Worksheet

Directions: Answer in the space provided. Be sure to read the question carefully.

- 1. Define matter
- 2. What is difference between a mixture and a pure substance? Give an example of each.
- 3. How can you tell and element apart from a compound?
- 4. What is an ion?
- 5. Complete the following sentences by filling in the appropriate word from the list below.

gas	plasma	physical
liquid	matter	chemical
solid	energy	Simms

- 1. ______ is anything that has mass and volume.
- 3. _____ changes alter the identity of a substance, whereas _____ changes do not.
- 4. Mr. ______ is amazing. The hair on his head is ridiculously thick and luscious.

Block:	
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6. Identify whether each of the following change is a physical change or a chemical change. Write "P" for physical and "C" for chemical

 water boiling	 iron rusting
 butter melting	 alcohol evaporating
 wood rotting	 baking a cake
 leaves changing colour	 glass breaking
 Milk sours	 mowing the lawn
 Evaporation	 Food is digested

- 7. What is the relationship between the kinetic energy (KE) of molecules and their physical state?
- 8. How can you determine whether a change in matter is physical or chemical?
- 9. Classify each of the following as an element (e), a compound (c) or a mixture (m).

 Gold	 Air
 Water	 Carbon Dioxide
 Seawater	 Silver
 Sugar	 Ice
 A chocolate sundae	 A cheeseburger

- 10. Classify each of the following as an intensive (I) or extensive (E) property
 - ______ Mass
 ______ Colour

 ______ Density
 ______ Volume

 ______ Melting Point
 ______ Length