

# NOMENCLATURE REVIEW

## Molecular Compounds, Ionic Compounds, & Acids

### ANSWER KEY

1. barium sulfite
2. ammonium phosphate
3. phosphorus pentabromide
4. magnesium sulfate
5. calcium oxide
6. phosphoric acid
7. sodium dichromate
8. magnesium oxide
9. sulfur trioxide
10. copper(II) nitrate
11. hydroiodic acid
12. dinitrogen monoxide
13. manganese(II) oxide
14. silver nitrate
15. diarsenic pentoxide
16. iron(III) oxide
17. hypochlorous acid
18. dinitrogen trioxide
19. hydrofluoric acid
20. oxalic acid
21. sodium bicarbonate
22. silicon tetrabromide
23. copper(II) chloride
24. nitrous acid
25. tin(IV) oxide
26. barium chromate
27. HBr
28.  $\text{Cr}_2(\text{CO}_3)_3$
29. MgS
30.  $\text{ICl}_3$
31. LiH
32.  $\text{NH}_4\text{OH}$
33.  $\text{CaCl}_2$
34.  $\text{H}_2\text{Se}$
35.  $\text{Fe}_3\text{N}_2$
36.  $\text{Al}(\text{OH})_3$
37.  $\text{SnF}_2$
38.  $\text{SCl}_4$
39.  $\text{HgI}_2$
40.  $\text{P}_2\text{O}_5$
41.  $\text{H}_2\text{SO}_3$
42.  $\text{Pb}(\text{NO}_3)_2$
43.  $\text{H}_2\text{O}$
44.  $\text{Na}_2\text{C}_2\text{O}_4$
45.  $\text{HClO}_4$
46.  $\text{HClO}_2$
47.  $\text{SiO}_2$
48.  $\text{H}_2\text{CO}_3$
49.  $\text{NaClO}_3$
50.  $\text{XeF}_6$
51.  $\text{Ni}(\text{NO}_3)_2$
52.  $\text{KClO}_4$