Chemistry 11 – Unit 2			Chemistry Math
Name	Block:	Date:	

Chemistry 11 **UNIT 2 REVIEW**

1. Complete the following table (try from memory first!)

Profiv

Abbreviation

Abbreviation	Exponent
μ	
	10^{6}
d	
	10 ⁻⁹
k	
	10 ⁻²
	μ d

- 2. Convert the following:
 - a. 0.00085L into microliters
 - b. 432 ng into grams
 - c. 50 ks into Ms
 - d. $2 cg into \mu g$
 - e. 12 pL into cL
 - f. 0.35 g/s into g/min
 - g. 70 kV into mV
 - h. 0.1 dm into millimetres
 - i. 3.46 mg/s into kg/ms
 - j. 0.96 kg/L into $\text{mg/}\mu\text{L}$

3. Of the following balances, which is the most **precise**?

How do you know? _

Balance A 0.00 g000.

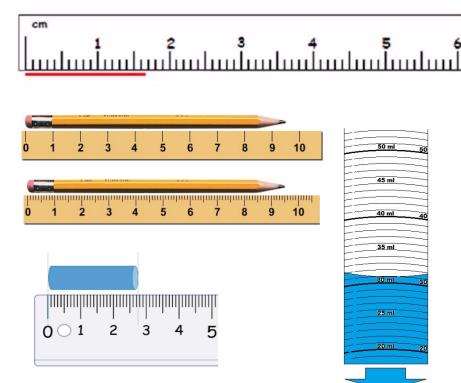


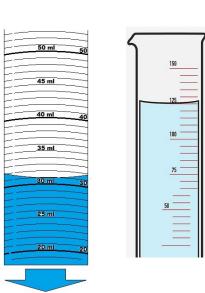


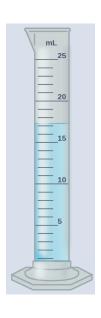
- 4. What is the uncertainty of Balance A? _____ (hint: look at the decimal places)
- 5. What is the uncertainty of Balance C? _____
- 6. The ______ digit in any measurement has some uncertainty.
- 7. The number of certain digits + 1 uncertain digit is called the number of

8. What is meant by the **accuracy** of a measurement? _____

9. Determine the correct reading on the following scales, including the uncertainty term. Assume all rulers are in cm and graduated cylinders are in mL.







10. Determine the number of **significant digits** (figures) in each of the following numbers.

a) 45.002 _____ b) 3.400 x 10⁻⁴ ____ c) 0.000003 ____ d) 3000 ____ 11. Express each of the following numbers to **2 significant digits** (figures).

a) 45670 _____ b) 0.00003448 ____ c) 3000000 ____

d) 23.0954 _____ e) 4.56219 x 10³ ____ f) 9 ____

12. Perform the following calculations and express the answer in the correct number of significant digits or decimal places as justified by the data. Do not forget the rules for multiplication and division and for addition and subtraction.

a) $3.4589 \times 0.0112 =$ _____ b) $(5.600 \times 10^{-7}) \div 0.700 =$ _____

c) 8.6 + 0.473 = _____ d) 3.2697 - 0.411 = _____

e) $(2.3 \times 10^{-7}) \times 8.22298 =$

f) $(5.2468 \times 0.923) + (3.00210 \times 1.9999) =$

g) $(6.210 + 0.92) \times (3.75411 + 1.32410) =$

h) (222.115 – 4.56892) = (32.98 – 25.22316)