

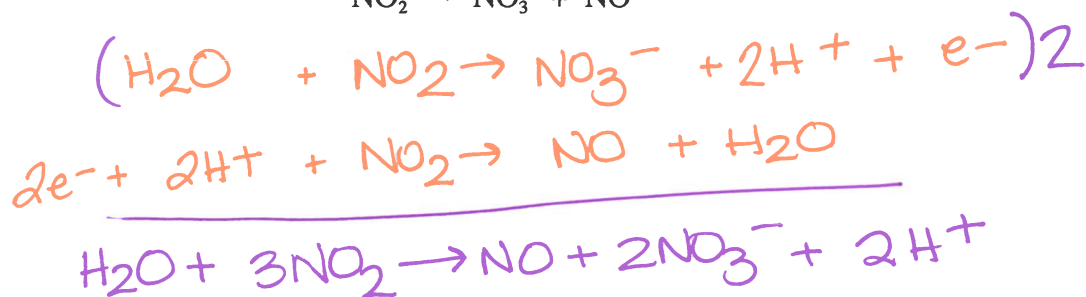
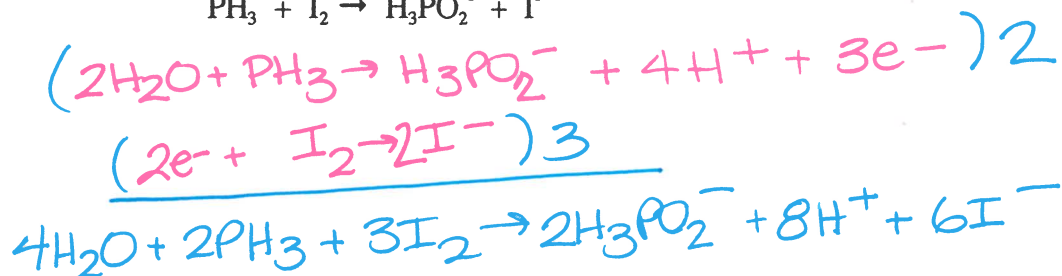
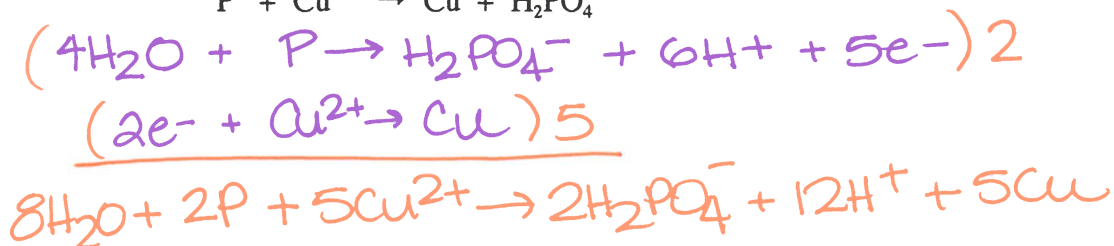
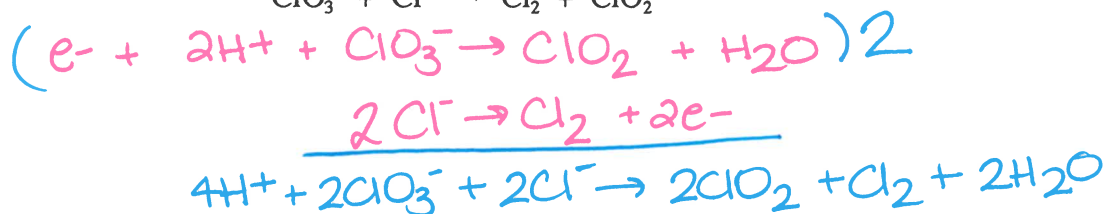
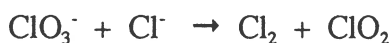
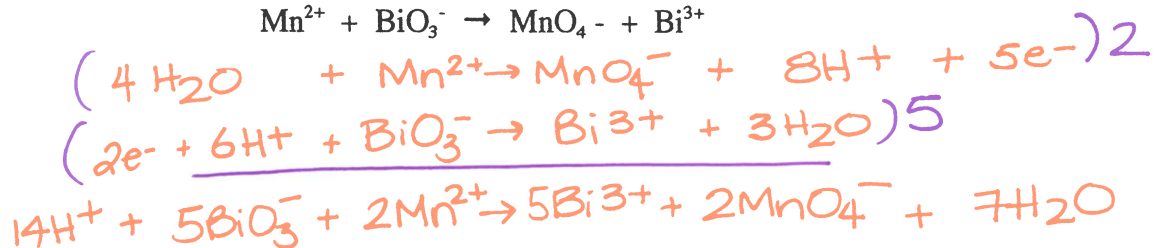
Name _____ Block: _____ Date: _____

Chemistry 12
BALANCING REDOX EQUATIONS

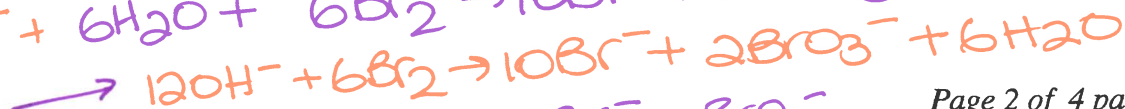
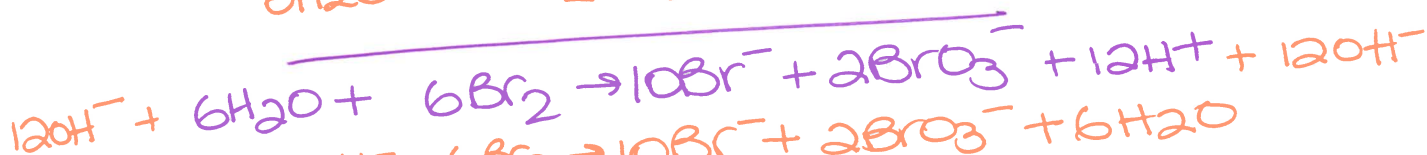
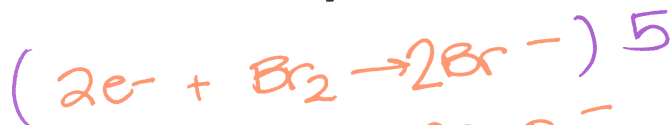
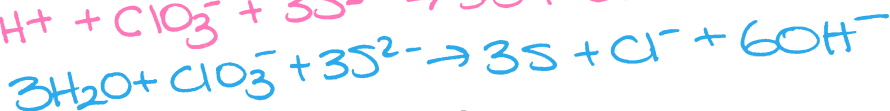
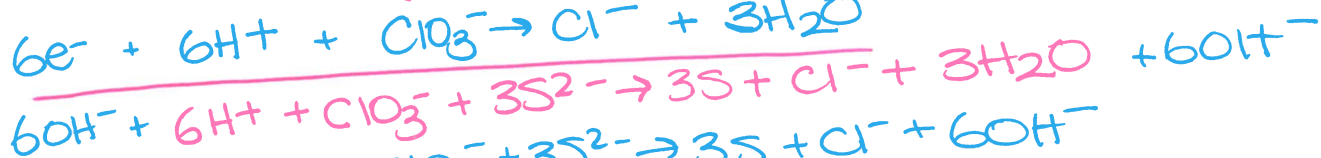
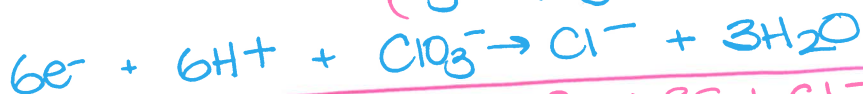
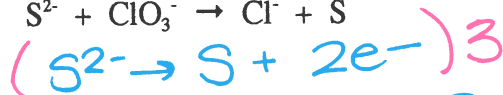
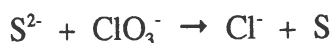
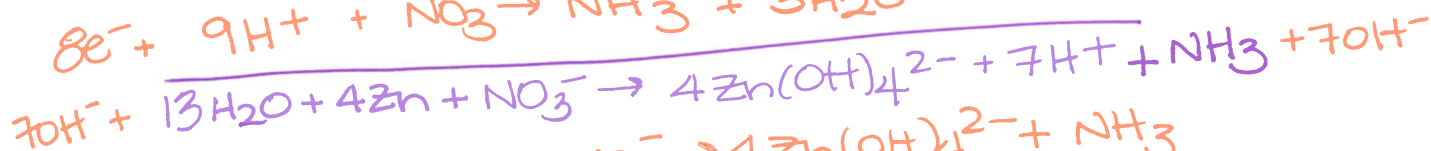
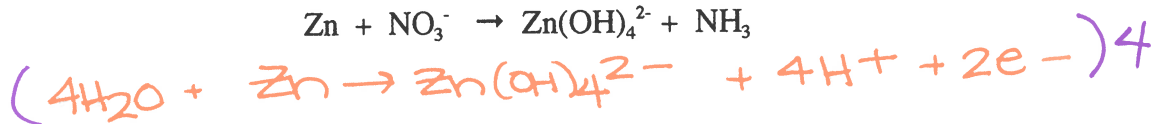
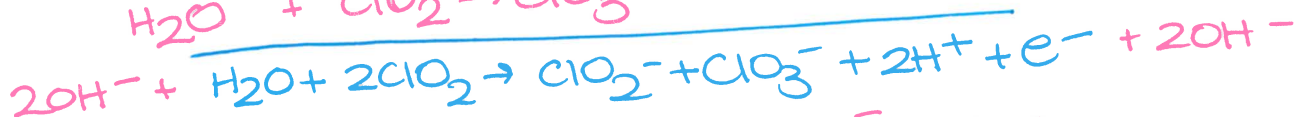
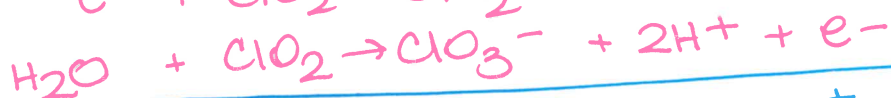
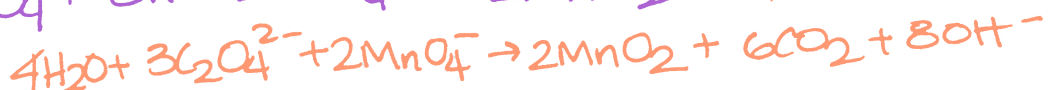
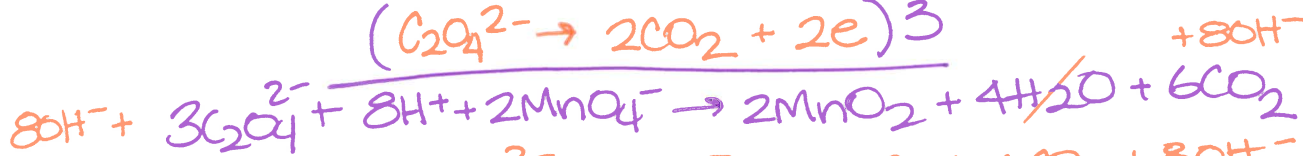
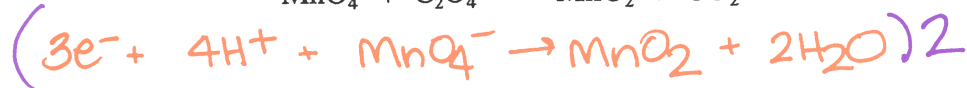
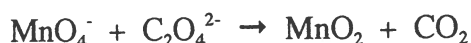
KEY

Balance the following redox equations using half-reactions.

A. ACID SOLUTIONS



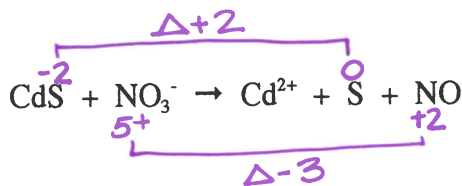
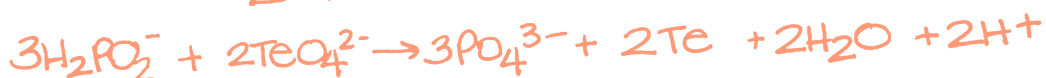
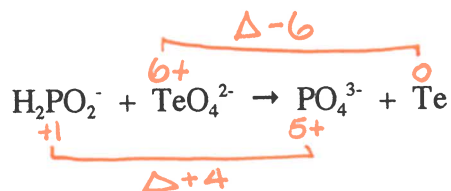
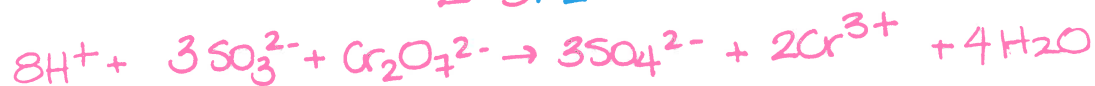
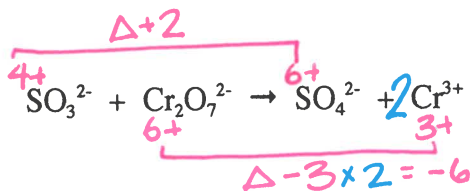
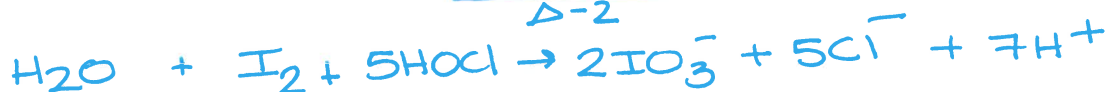
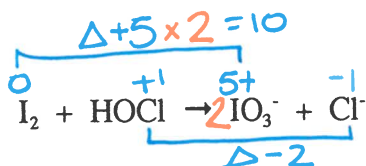
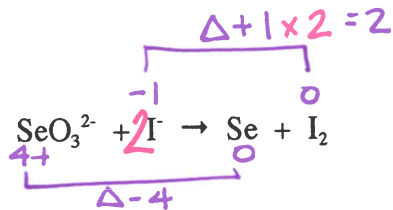
B. BASIC SOLUTIONS



simplify
÷2

Balance the following using oxidation numbers.

A. ACID SOLUTIONS



B. BASIC SOLUTIONS

